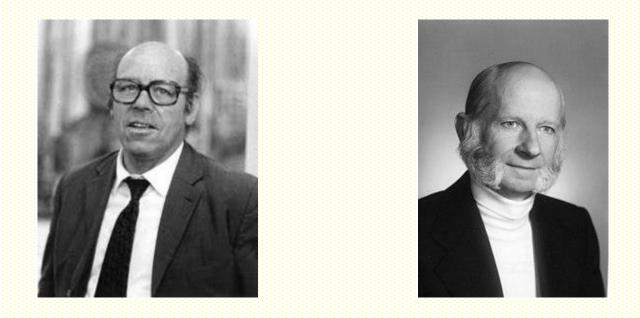


# Trapping lons and Cooling Atoms

**IPHO 2024-IRAN** 



## A. Paul Trap



The Nobel Prize in Physics 1989 was divided, .... the other half jointly to Hans G. Dehmelt and Wolfgang Paul "for the development of the ion trap technique". From nobelprize.org



## How to trap an Ion?

#### Earnshaw's theorem

a collection of point charges cannot be maintained in a stable stationary equilibrium configuration solely by the electrostatic interaction of the charges.

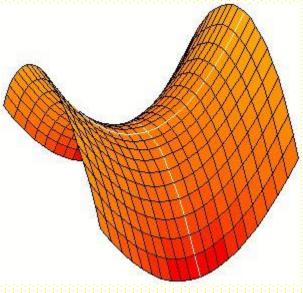
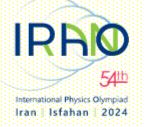
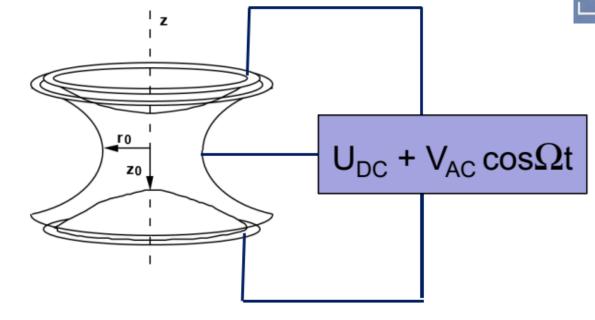


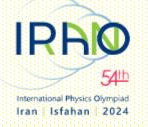
Photo from wikipedia



#### Paul Trap: Using Dynamical Stability



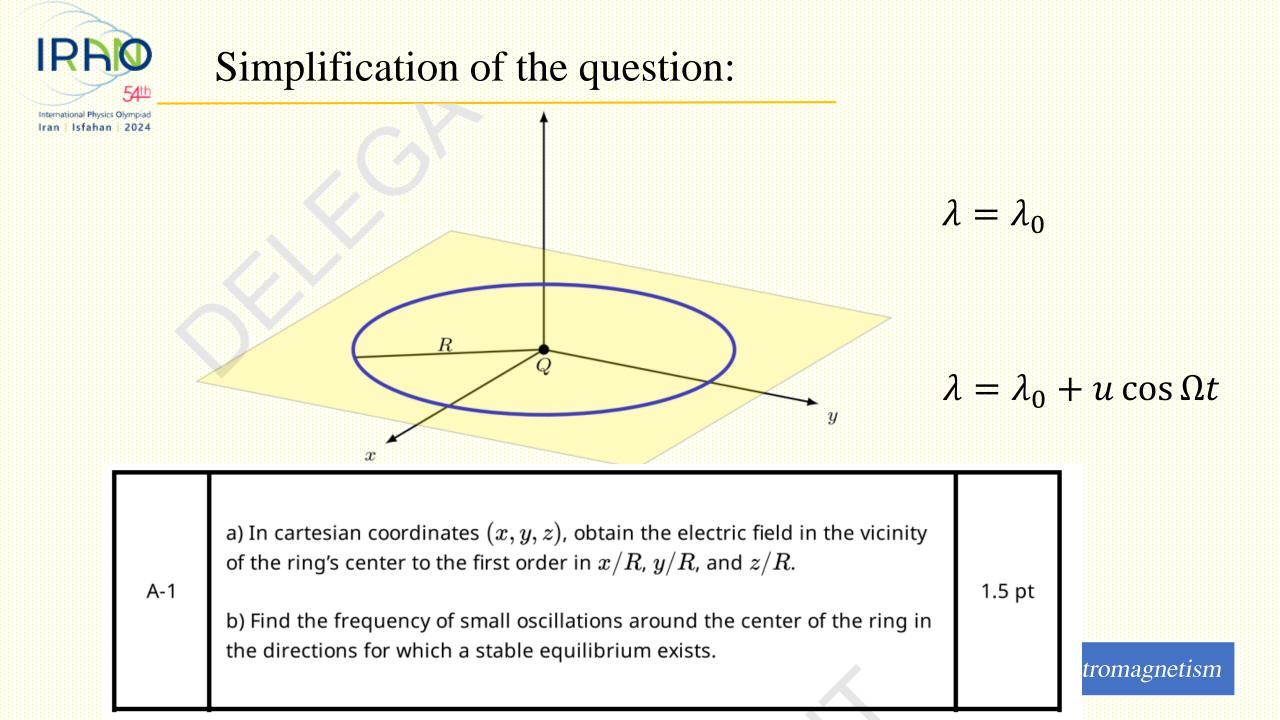










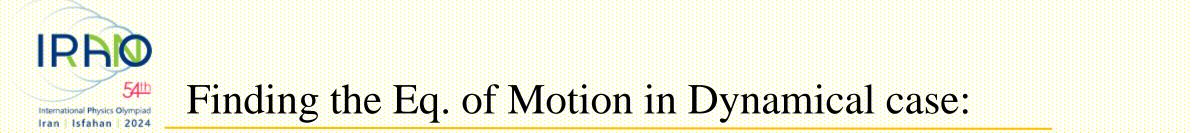




#### The ideas needed to solve:

- 1) Writing Coulomb's Law and very simple integrating.
- 2) Using suitable approximations and expansions.
- 3) Using Gauss's Law.

$$\vec{E}(x, y, x) \simeq \frac{\lambda}{4\epsilon_0 R^2}(-x, -y, 2z)$$

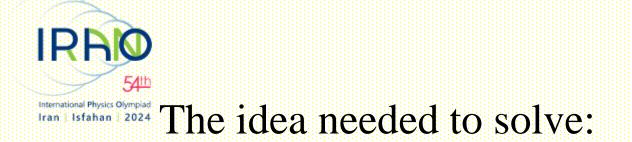


$$\lambda = \lambda_0 + u \cos \Omega t \longrightarrow \ddot{z} = (+k^2 + a \Omega^2 \cos \Omega t) z$$

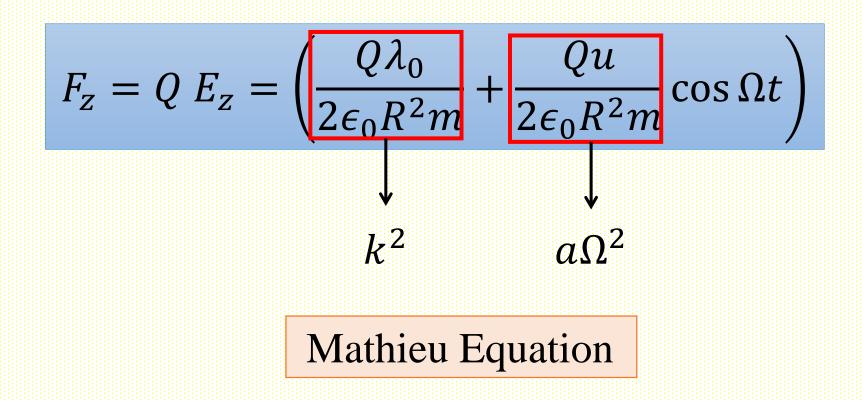
A-2 Write *a* and *k* in terms of the known parameters.

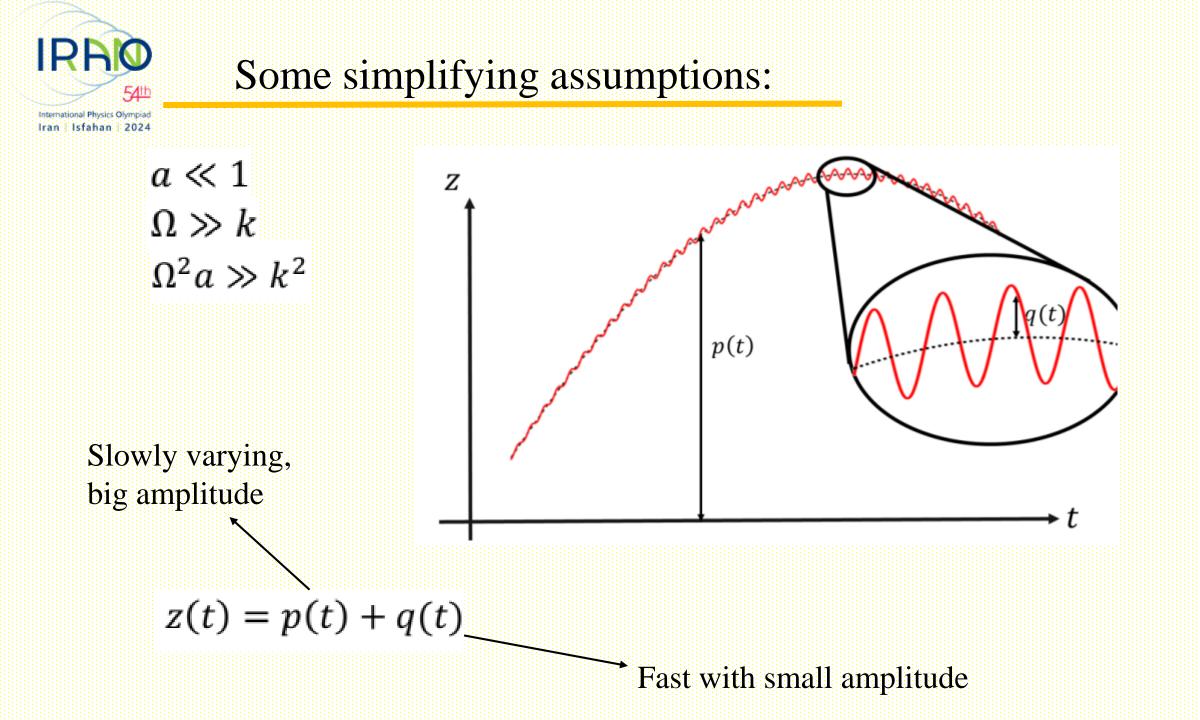
0.6 pt

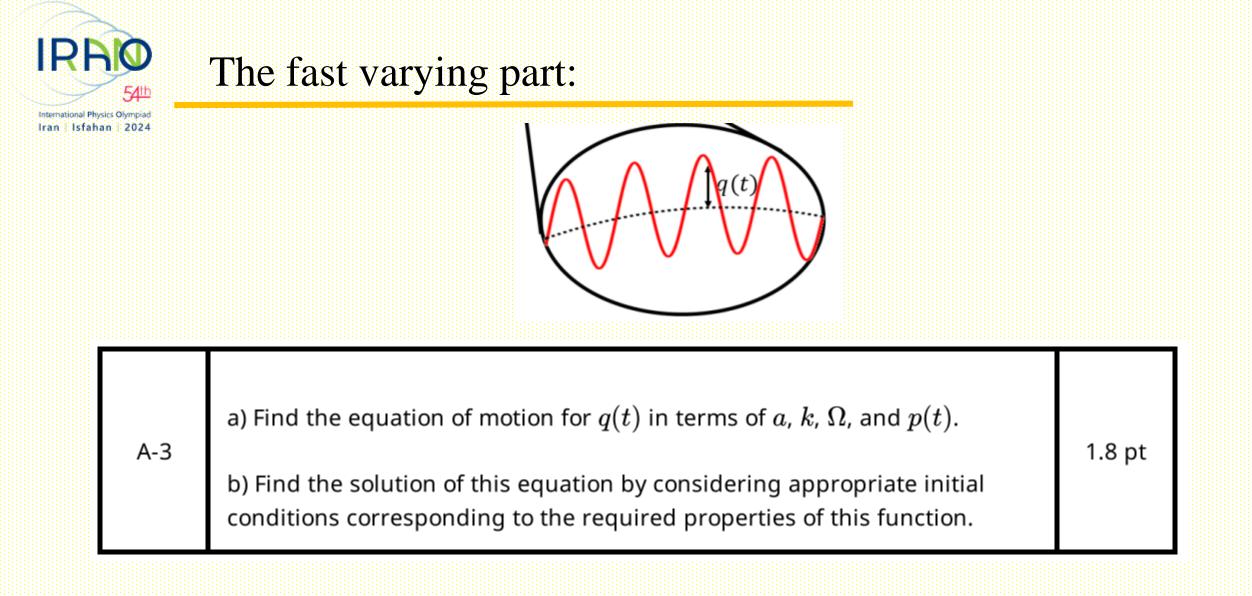
Syllabus: Mechanics, Oscillations and waves



Writing Newton's Second Law







Syllabus: Mechanics, Oscillations and waves

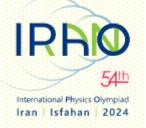


#### The ideas needed to solve:

p is nearly constant  $q \ll p$  $\Omega^2 a \gg k^2$ 

 $\ddot{p} + \ddot{q} = (k^2 + a\Omega^2 \cos \Omega t)(p + q)$ 

 $\ddot{q} = a\Omega^2 \cos \Omega t p$ 



 $\ddot{q} = a\Omega^2 \cos \Omega t p$ 

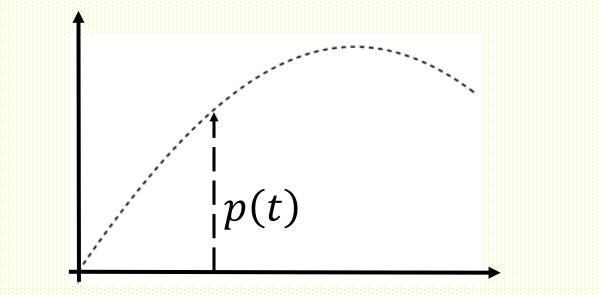
#### p is nearly constant

$$q = -a p \cos \Omega t + c_1 t + c_2$$

#### q should remain small with zero mean

#### The slow varying part:

A-4



a) Consider the mean effect of the rapidly varying component and obtain an effective equation of motion for p(t).

1.5 pt

b) Investigate the stability of the equilibrium point and find the condition for a stable equilibrium.

Syllabus: Mechanics, Oscillations and waves



#### The ideas needed to solve:

1) Substituting the answer for q(t) in the original Eq. of motion.

2) Using the mean values  $\langle \cos \Omega t \rangle = 0$  and  $\langle \cos^2 \Omega t \rangle = 1/2$ 

 $\ddot{p} \neq \left(k^{2} + \frac{a^{2} \Omega^{2} p}{2}\right) post \Omega t ) (p \Omega^{2} p) cos^{2} \Omega t$ 

Stability Condition:  $\Omega > \sqrt{2} \frac{k}{a}$ 

### **IRNO** Numerical estimate of the frequency needed

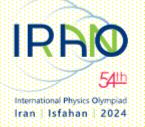
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54"

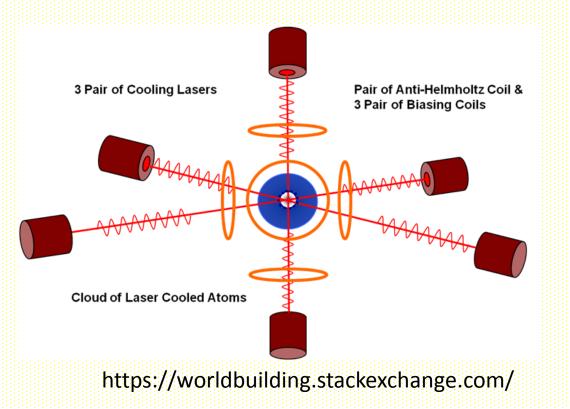


Assume that  $\lambda_0 = 8 \times 10^{-9} \text{ C/m}$  and R = 10 cm. We would like to use this device to trap a singly ionized atom 100 times heavier than a hydrogen atom.

 $k = 2 \times 10^5 \text{ rad/s}$   $\Omega_{\min} \simeq 7 \times 10^6 \text{ rad/s}$ 



## B. Cooling Atoms



The same concept in IPhO 2009 Mexico, yet very different approach.



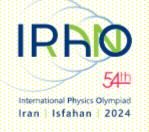
#### Laser Cooling

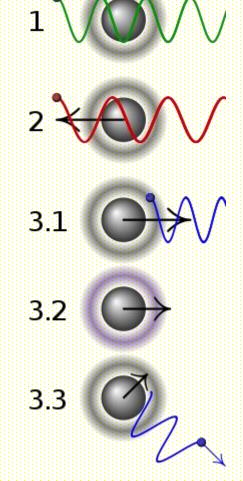
Photo from the Nobel Foundation archive. Steven Chu Prize share: 1/3 Photo from the Nobel Foundation archive. Claude Cohen-Tannoudji Prize share: 1/3



Photo from the Nobel Foundation archive. William D. Phillips Prize share: 1/3

The Nobel Prize in Physics 1997 was awarded jointly to Steven Chu, Claude Cohen-Tannoudji and William D. Phillips "for development of methods to cool and trap atoms with laser light" From nobelprize.org

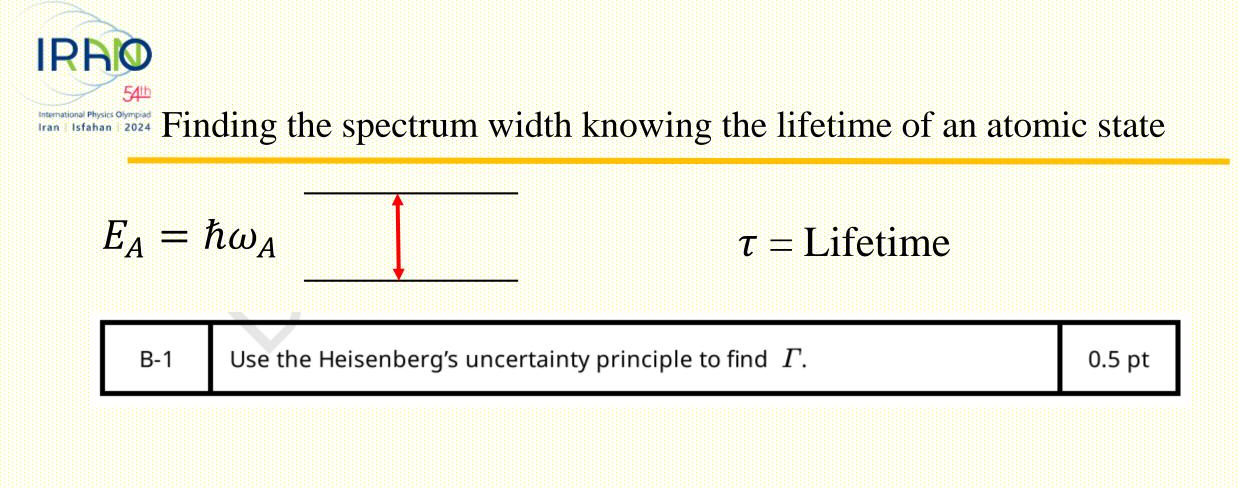




The atom absorbs mainly the blueshifted lights due to Doppler effect.

The mechanism effectively reduces the energy of the atom

Photo from wikipedia



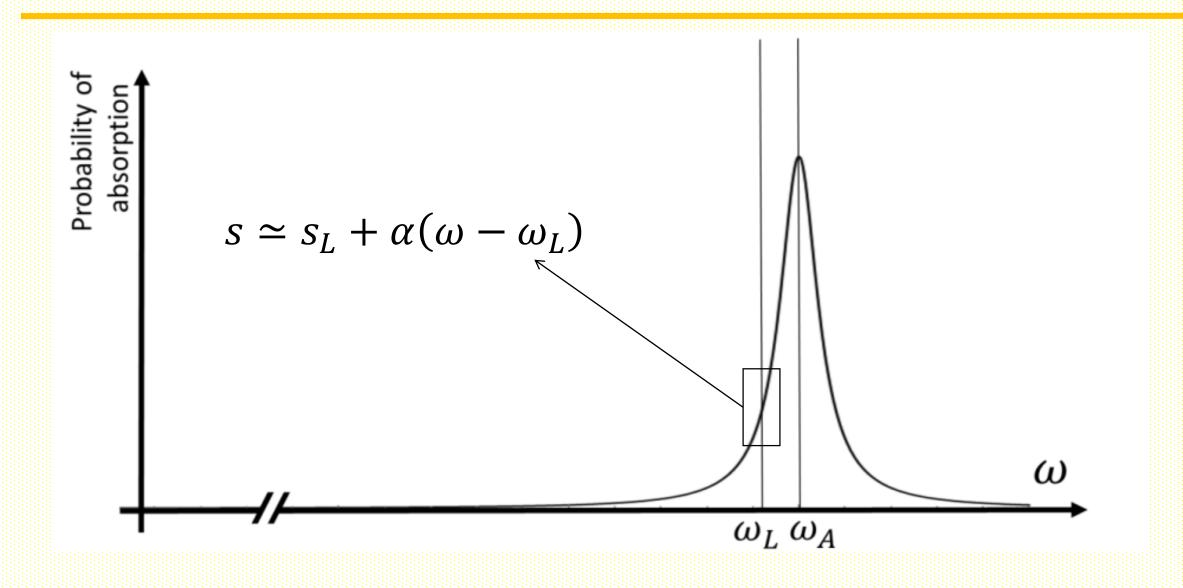
$$\Delta E \times \Delta t \simeq \hbar \quad \longrightarrow \quad \Gamma = \Delta \omega \simeq 1/\tau$$

Syllabus: Quantum Physics



### 54 The probability of absorption in the atom's reference frame

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#### The changes to absorption rate due to the Doppler effect and the effective force on the atoms

a) Assume that the trapped atom is moving with a velocity,  $v = v_x$  in the lab frame. In the frame of reference of the atom, calculate the collision rate of the photons, incident from each of the two directions, with the B-2 atoms (denoted by  $s_+$  and  $s_-$ ) and the rate of absorption of momentum in each direction (denoted by  $\pi_+$  and  $\pi_-$ ). b) Determine the effective force on the atom as a function of v,  $k_{\rm L} = \omega_{\rm L}/c$ ,

 $\hbar$ , and lpha, in the reference frame of the laboratory. Assume  $s_{
m L} \ll lpha \omega_{
m L}$ ,

Syllabus: Mechanics, Relativity



In the reference frame of the atom:

$$\omega_{+} = \omega_{L} \left( 1 + \frac{\nu}{c} \right)$$
$$\omega_{-} = \omega_{L} \left( 1 - \frac{\nu}{c} \right)$$

$$s_{+} = s_{L} + \alpha \omega_{L} \frac{\nu}{c}$$
$$s_{-} = s_{L} - \alpha \omega_{L} \frac{\nu}{c}$$



The rate of momentum absorption:

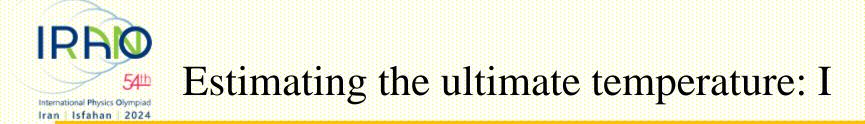
$$\pi_{+} = s_{+} \times (-\hbar k_{+})$$

$$\pi_{-} = s_{-} \times (+\hbar k_{-})$$

$$\pi_{+} + \pi_{-} = -2\hbar k_{L} \alpha \omega_{L} \frac{v}{c} \left(1 + \frac{s_{L}}{\alpha \omega_{L}}\right)$$

$$F = -(2\alpha\hbar k_L^2)v$$

The same force in the lab's frame up to the order v/c



#### Finding the energy absorption rate of a completely stopped atom

B-3

Considering the momentum of the atom after such a process for the two possible outcomes, calculate the average power absorbed by the atom.

1.0 pt

Syllabus: Mechanics



Before absorbing the photon

$$p_0 = 0$$

After emitting a photon:

After absorbing the photon:

 $p_1 = \hbar k_L$ 

$$E_f \rangle = \frac{1}{2} \times 0 + \frac{1}{2} \times \frac{(2\hbar k_L)^2}{2m} = \frac{\hbar^2 k_L^2}{m}$$

Input power = 
$$\frac{\hbar^2 k_L^2}{m\tau}$$

 $p_{f2} = 2\hbar k_L$ 

 $p_{f1} = 0$ 



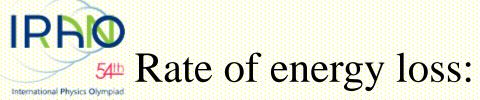
Finding the energy dissipation rate of an atom and the steady state temperature

B-4

Consider the force calculated in Task B-2 and calculate the output power. Then, calculate the average value of  $v^2$  at equilibrium. Using your knowledge of the kinetic theory of gases estimate the temperature of the atoms.

0.8 pt

Syllabus: Mechanics, Thermodynamics and statistical physics



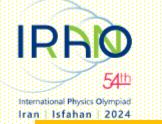
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$$P_{\rm out} = F.\, v = -2 \,\alpha \hbar k_L^2 v^2$$

Steady state condition:

$$P_{\rm out} + P_{\rm in} = 0 \longrightarrow \overline{v^2} = \frac{\hbar\Gamma}{2\alpha m}$$

$$T = \frac{\hbar\Gamma}{2\alpha k_B}$$



#### Numerical estimate of the ultimate temperature:

B-5	Estimate this temperature, for an atom 100 times heavier than a hydrogen	0.4 pt
	atom. Assume that $\omega_{ m L}=2 imes10^{16}{ m rad}/{ m s}$ , $ au=5 imes10^{-9}{ m s}$ , and $lpha=4.$	

$$T = 2 \times 10^{-4} K$$



